

2-(DIPHENYLPHOSPHINOYL) PYRROLIDINES;
VERSATILE REAGENTS FOR THE SYNTHESIS OF HETEROCYCLIC ENAMINES AND ENAMIDES.

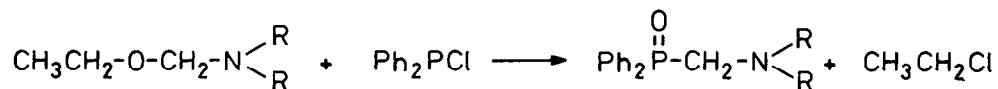
B.H. Bakker, D.S. Tjin A-Lim, and A. van der Gen*
Gorlaeus Laboratories, Department of Organic Chemistry, University of Leiden,
P.O. Box 9502, 2300 RA Leiden, The Netherlands

Summary: A method is described for the preparation of N-substituted 2-(diphenylphosphinoyl) pyrrolidines. Application of these phosphine oxides in the Horner-Wittig reaction affords heterocyclic enamines and enamides in good yields.

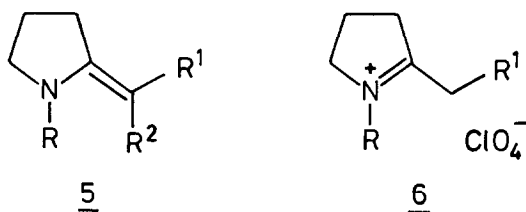
As part of our continuing interest in the synthetic applications of α -heterosubstituted diphenylphosphine oxides,¹⁻⁴ we are investigating the possibility of synthesizing heterocyclic enamines and enamides, versatile intermediates in natural product synthesis,^{5,6} by the Horner-Wittig reaction. A recent publication on the synthesis and application of N-carbaethoxy-2-(diphenylphosphinoyl) piperidine⁷ prompts us to report our results in the pyrrolidine series.

Frequently, diphenylphosphine oxides show superior properties in the (Horner)-Wittig reaction compared to phosphonium salts and phosphonates.⁸ This is well demonstrated by α -aminomethyl diphenylphosphine oxides, which are versatile reagents for the synthesis of enamines,^{2,3} whereas, in our experience, the corresponding phosphonium salts and phosphonates are not.^{2,9}

These α -aminomethyl diphenylphosphine oxides are commonly prepared by an Arbusov reaction of N-ethoxymethylamines with chlorodiphenylphosphine:



Application of this reaction scheme to N-substituted tetrahydro-1,3-oxazines 1 should give rise to ring opened aminomethyldiphenylphosphine oxides 2, bearing a chloro substituent at the right position for nucleophilic ring closure to a pyrrolidine derivative.

TABLE I : Conversion of carbonyl compounds R^1R^2CO into heterocyclic enamines and enamides.

Entry	R	R ¹	R ²	Yield(%) ^a	m.p. (°C) ^b
1	CH ₃	Ph	H	81	108 - 109 ^c
2	CH ₃	p-MeOC ₆ H ₄	H	57	87 - 88
3	CH ₃	n-C ₃ H ₇	H	65	113 - 115 ^d
4	CH ₃	CH ₃	H	50	203 - 205 ^e
5	CH ₂ Ph	Ph	H	79	195 - 196
6	CH ₂ Ph	p-MeOC ₆ H ₄	H	76	170 - 172
7	COPh	Ph	H	84 ^g	143 - 145 ^f
8	COPh	C ₂ H ₅	H	87 ^h	oil
9	COPh	-CH ₂ CH ₂ CH(CH ₂ CH ₂ -) C(CH ₃) ₃		67	115 - 116
10	COPh	Ph	Ph	78	176 - 178

a. For the enamines (entries 1 to 6), yields and m.p. refer to isolated perchlorates 6.

b. m.p. are uncorrected

c. Lit.¹¹ : m.p. 109-110 °C

d. Lit.¹² : m.p. 117-118 °C

e. Lit.¹² : m.p. 208-209 °C

f. m.p. of E-isomer, obtained by recrystallization from hexane-CH₂Cl₂.

g. E/Z ratio 5:2.

h. E/Z ratio 1:1.

The N-benzoyl substituted anion 4c is strongly stabilized by dipole interactions,¹³ even so a fast reaction was observed with aldehydes at -70 °C. Completion of the Horner-Wittig reaction was easily effected by stirring overnight at room temperature. The resulting enamides are stable compounds and are obtained in good yields as mixtures of E- and Z-isomers (entries 7 and 8).¹¹ The increased stability of anion 4c allows reaction with carbonyl compounds at

higher temperatures and thereby extension of the reaction to ketones. 4-t-Butylcyclohexanone and benzophenone did not add to the anion at -70°C , but prolonged reaction times at room temperature (48 hrs) afforded the enamides, which were obtained in yields of 67% and 78% respectively, after crystallization (entries 9 and 10).

Application of the novel route to heterocyclic phosphine oxides described in this communication to the synthesis of other N- and O-heterocyclic systems is currently being investigated and will be reported separately.

REFERENCES

1. C.G. Kruse, E.K. Poels and A. van der Gen, *J.Org.Chem.* 44, 2911 (1979).
2. N.L.J.M. Broekhof, F.L. Jonkers and A. van der Gen, *Tetrahedron Letters* 1979, 2433.
3. N.L.J.M. Broekhof, F.L. Jonkers and A. van der Gen, *Tetrahedron Letters* 21, 2671 (1980).
4. T.A.M. van Schaik, A.V. Henzen and A. van der Gen, *Tetrahedron Letters* 24, 1303 (1983).
5. a. A.G. Cook, "Enamines: Synthesis, Structure and Reactions", Marcel Dekker Ed. New York 1969, Chapter 7.
b. K. Blaha and O. Cervinka, *Adv. Heterocyclic Chem.* 6, 147 (1963).
6. a. G.R. Lenz, *Synthesis* 1978, 489.
b. I. Ninomiya, *Heterocycles* 14, 1567 (1980); *idem, ibid.* 15, 1433 (1981).
7. T. Shono, Y. Matsumura and T. Kanazawa, *Tetrahedron Letters* 24, 4577 (1983).
8. J.I. Grayson and S. Warren, *J.C.S. Perkin I*, 1977, 2263.
9. N.L.J.M. Broekhof, Thesis, University of Leiden 1982.
10. a. Z. Eckstein and T. Urbansky, *Adv. Heterocyclic Chem.* 2, 311 (1963).
b. H. Booth and R.U. Lemieux, *Can. J. Chem.* 49, 777 (1971).
c. A.R. Katritzky, V.J. Baker and F.M.S. Brito-Palma, *J.C.S. Perkin II*, 1980, 1739.
11. Satisfactory analytical data were obtained for all new compounds.
12. R. Lukes, *Chem. Listy* 27, 392 (1933).
13. R. Lukes, V. Dedik and L. Novotny, *Chem. Listy* 52, 654 (1958).
14. P. Beak and W.J. Zajdel, *J.Am.Chem.Soc.* 106, 1011 (1984).

ACKNOWLEDGEMENT

The investigations were supported by the Netherlands Foundation for Chemical Research (S.O.N.) with financial aid from the Netherlands Organization for the Advancement of Pure Research (Z.W.O.).

(Received in UK 22 May 1984)